Part 1 – Multiple Choice (30 points)

1. In this list, which substance can be classified as a chemical?
	1. sleep
	2. heat
	3. cold
	4. salt
	5. temperature
2. The first step in the scientific method is \_\_\_\_\_\_\_\_.
	1. using technology
	2. making observations
	3. forming a hypothesis
	4. doing experiments
	5. proposing a theory
3. One way to enhance your learning in chemistry is to \_\_\_\_\_\_\_\_.
	1. study a little every day
	2. form a study group
	3. go to office hours
	4. be an active learner
	5. all the above
4. The amount of space occupied by a substance is its \_\_\_\_\_\_\_\_.
	1. mass
	2. volume
	3. density
	4. weight
	5. length
5. Which of the following numbers is the smallest?
	1. 4.0 × 10-12
	2. 4.0 × 10-6
	3. 4.0 × 10-8
	4. 4.0 × 10-2
	5. 4.0 × 1015
6. Which of the following is the largest unit?
	1. millimeter
	2. micrometer
	3. meter
	4. decimeter
	5. kilometer
7. The cubic centimeter (cm3 or cc) has the same volume as a \_\_\_\_\_\_\_\_.
	1. cubic inch
	2. cubic liter
	3. milliliter
	4. centimeter
	5. cubic decimeter
8. Compounds are pure substances that by definition consist of \_\_\_\_\_\_\_\_.
	1. a single element
	2. oxygen and hydrogen
	3. solids
	4. two or more elements in combination
	5. gases
9. When gold is melted and formed in a mold to make a piece of jewelry, what type of change is taking place?
	1. a physical change
	2. a chemical change
	3. a change of size
	4. evaporation
	5. boiling
10. When you observe the formation of fog on a cool, humid day, what type of event are you observing?
	1. a chemical change in oxygen
	2. a physical change in air
	3. a physical change in water
	4. a chemical change in water
	5. a combination of nitrogen and oxygen
11. The dietary calorie (Cal) is equal to \_\_\_\_\_\_\_\_.
	1. 1 000 kilocalories
	2. 100 kilocalories
	3. 100 calories
	4. 10 calories
	5. 1 kilocalo

Part 2 – 70 points (54)

1. (4 points) Give the length of the stick to the correct number of significant figures.



1. ( 10 points) Perform the appropriate action on each of the following numbers or calculations
	1. Round 93.592299 to three significant figures.
	2. How many significant figures are in 0.00042010?
	3. Write the number 29508132502 in scientific notation with 4 significant figures.
	4. Perform the following calculations to the correct number of significant figures.

$$432.811 cm+729.3 cm= $$

$$35.342 g PbO\_{2}×\frac{1 mol PbO\_{2}}{239.1 g PbO\_{2}}×\overset{this is exact}{\overbrace{\frac{2 mol O}{1 mol PbO\_{2}}}}×\frac{16.0 g O}{1 mol O}=$$

1. (5 points) In England, a person is weighed in stones. If one stone is 14.0 lb, what is the mass, in kilograms, of a person who weighs 12.4 stones?
2. (5 points) In a candy factory, the nutty chocolate bars contain 32.0% by mass pecans. If 12.8 kg of pecans were used for candy last Tuesday, how many pounds of nutty chocolate bars were made?
3. (5 points) A gem has a mass of 5.62 g. When the gem is placed in a graduated cylinder containing 5.00 mL of water, the water level rises to 6.34 mL. What is the density of the gem?
4. (5 points) A wooden sculpture has a density of 1.21 g/mL. If the sculpture has a mass of 6.47 kg, what is the volume of the sculpture in gallons? (1 gallon = 4 quarts)
5. (5 points) A car travels at 55 miles per hour and gets 13.5 km per liter of gasoline. How many liters of gasoline are needed for a 6.00 hour trip?
6. (10 points) Classify each of the following substances as an element, a compound, a mixture. (Check the appropriate box for each substance.)

|  |  |  |  |
| --- | --- | --- | --- |
| Substance | Element | Compound | Mixture |
| A soft drink |  |  |  |
| Helium gas (He) |  |  |  |
| Methane (CH4) in natural gas |  |  |  |
| Ice (H2O) |  |  |  |
| Bronze (an alloy of Cu and Sn) |  |  |  |

1. (5 points) A German cookie recipe calls for a cooking temperature of 145oC. What is this temperature in oF?

|  |  |
| --- | --- |
| protein | 4 kcal |
| Fat | 9 kcal |
| carbohydrate | 4 kcal |

1. (5 points) A chocolate chip cookie contains 2.0 g of protein, 30.0 g of carbohydrate, and 10.0 g of fat. Using the table on the right, determine the number of kcal in that chocolate chip cookie. (Ignore significant figures here and calculate value to the nearest kcal.)
2. (10 points) Give an example of each of the following:

|  |  |
| --- | --- |
| * 1. A metallic element
 |  |
| * 1. A transition metal
 |  |
| * 1. A halogen
 |  |
| * 1. A noble gas
 |  |
| * 1. An alkaline earth
 |  |